

The ionization parameter

- ◆ U, the ratio of ionizing photon to hydrogen densities
- ◆ See **Davidson & Netzer 1979**

2.7 Photoionization of Heavy Elements

Finally, let us examine the ionization of the heavy elements, of which O, C, Ne, N, Si, and Fe, with abundances (by number) of order 10^{-3} to 10^{-4} that of H, are the most abundant. The ionization-equilibrium equation for any two successive stages of ionization i and $i + 1$ of any element X may be written

$$n(X^{i+1}) \int_0^\infty \frac{4\pi J_\nu}{h\nu} \sigma_{\nu,0}(X^{i+1}) d\nu = n(X^i) \Gamma(X^i) \quad (2.30)$$

$$= n(X^{i+1}) n_e \sigma_{\nu,0}(X^i, T),$$

where $n(X^i)$ and $n(X^{i+1})$ are the number densities of the two successive stages of ionization; $\sigma_{\nu,0}(X^{i+1})$ is the photoionization cross section from the ground level of X^i

U and T(star) determine ionization

- ◆ No matter how intense the radiation field, how large the U, ions with ionization potentials higher than the highest energy in the SED cannot be produced

